

From ref. 1

#### INCORPORATING POLYMERS FOR REDUCED

#### COST OF SMART WINDOWS

A presentation by Team Smart Cicada

May 14, 2014

# MOTIVATION AND PURPOSE

- Low eco-impact buildings
- Smart Window a solid state device that changes based on applied voltage
- Switchable Mirror goes from reflective to transparent for enhanced efficiency
- Replace ITO and tungsten oxide with PEDOT:PSS and PANI



COMSOL simulation of optical transmittance

# **DESIGN AND GOALS**

- Switchable Mirror Move hydrogen into and out of an active layer to cause changes in optical properties
- Electrode/Ion storage/Electrolyte/Active Layer/Electrode
- Tajima's group device had a layering of:
  - ITO/WO<sub>3</sub>/PEI/Pt-Mg<sub>4</sub>Ni/ITO
  - Our device replaces ITO with PEDOT:PSS as a conductive transparent electrode and WO<sub>3</sub> with PANI as the ion storage layer:
  - PEDOT:PSS/PANI/PEI/Pt-Mg<sub>4</sub>Ni/PEDOT:PSS



• GOALS

# TECHNICAL APPROACH - POLYMER PROCESSING

- Shape, thickness, and structure of each layer is dependent on processing conditions
  - Polymer Synthesis
    - PANI
    - PEI
    - PEDOT:PSS
  - Deposition Methods
    - Spin Coating
    - Meyer Rod Coating



# TECHNICAL APPROACH - SPUTTERING PROCESSING

- Sputtering Mg4Ni & Pt
  - Mg4Ni layer on PEDOT:PSS
    - Goal of 70nm of Mg4Ni
    - Power ratio of 1.88 : 1 for Mg : Ni
    - Absorbing and desorbing hydrogen
  - Pt layer on Mg4Ni layer
    - Goal of 4nm of Pt
    - Shiny and like mirror





AJA Sputtering unit



• Modelling through hydrogen diffusion: Fick's First Law

$$J_{i} = -D_{i} \frac{\partial V}{\partial x} = D_{i} \cdot E_{i}$$

$$\int_{i} = \frac{\sigma_{i} \cdot V_{app} \cdot \varepsilon_{r,i}}{d_{total} \cdot \Im}$$

$$t_{i} = \frac{d_{i} \cdot \rho_{i} \cdot n_{i}}{J_{i} m_{w,i}}$$

$$t_{swtiching} = \frac{d_{total} \cdot \Im}{V_{app}} \sum_{i=1}^{n_{layers}} \frac{d_{i} \cdot \rho_{i} \cdot n_{i}}{\sigma_{i} \cdot \varepsilon_{r,i} \cdot m_{w,i}}$$





• Mass balancing to determine PANI thickness

$$\begin{split} N_{H,MgH_{2}/MgNiH_{4}} &= \frac{d_{MgH_{2}/MgNiH_{4}} \cdot A_{device} \cdot \rho_{MgH_{2}/MgNiH_{4}} \cdot m_{MgH_{2}/MgNiH_{4}} \cdot N_{A}}{m_{w,MgH_{2}/MgNiH_{4}}} \\ & \left| N_{H,PANI} = \frac{d_{PANI} \cdot A_{device} \cdot \rho_{PANI} \cdot m_{PANI} \cdot N_{A}}{m_{w,PANI}} \right| \\ d_{PANI} &= \frac{1.31 \cdot d_{Mg_{4}Ni} \cdot \rho_{MgH_{2}/MgNiH_{4}} \cdot m_{MgH_{2}/MgNiH_{4}} \cdot m_{w,PANI}}{m_{w,MgH_{2}/MgNiH_{4}} \cdot \rho_{PANI} \cdot m_{PANI}} \end{split}$$

# **ELECTRICAL PREDICTIONS**

- Make assumptions:
  - One layer is rate limiting
  - Most likely MgH2/MgNiH4 or PANI
  - Assume dimensions for our prototype and dielectric
  - Use data from literature to assess conductivity
- MgH2/MgNiH4 conductivity: ~1.32\*10^-8 /ohm-m
- PANI conductivity: ~3.8\*10^-8 /ohm-m





- Aimed to model the optical properties of our device through COMSOL Multiphysics 4.4
- Obtained a floating license through Dr. Phaneuf
  - Included Wave Optics Module
- Original Plan Fresnel Equations
  - Model transmittance and reflectance vs. wavelength/frequency
  - Could do this for a simple 2-layer, 3-D model
    - More advanced models proved to be difficult
    - Computing issues / Frequency sweep issues
- Final Simulations Maxwell's equations
  - Model transmitted light beam intensity through our device
  - Simple 2-D model with accurate layer thicknesses
  - Frequency Domain, time independent, FEM



- Performed multiple thought-experiments to test the validity of our assumptions and choices in COMSOL.
- Needed to estimate multiple material layers since we could not get sufficient experimental constants
- Applied necessary boundary conditions (transition, scattering)
- Modelled roughness at each interface with effective medium theory:.

Effective Medium Approximation	Equation				
Drude	$n_{eff}^2 = (1-\phi)n_c^2 + \phi n_d^2$				

#### **COMSOL 4.4 RESULTS**



### **COMSOL 4.4 RESULTS**

• Electric field of final design at 600 nm (left) and 900 nm (right).



# COMSOL 4.4 RESULTS

- Power out/in curve of our final design at 900 nm (right).
- Point plot of power out of each interface along the using a 600 nm plane wave (left).





# **POLYMER SYNTHESIS**



From left to right: concentrated PANI, PEDOT:PSS, and PEI solutions

# **PEDOT:PSS**

Preliminary Spin Coating Attempts



Uneven Coating

Final Deposition Techniques



Delicate Film

Multiple Layers Split



Preliminary Spin Coating Attempts



Rough, Uneven Coating, Agglomerations

Final Deposition Techniques



Too thin vs. Too rough, too thick

# THIN FILM RESULTS





- Measuring thickness of layers.
  - PANI layer by spin-coating 70nm
  - PANI layer by drop-casting 4um
  - PEDOT layer 70 nm
  - PEI layer 70nm
  - Mg4Ni layer & Pt layer 200nm



Profilometer Tenco Alpha Step 200

# **CHARACTERIZATION - PROFILOMETER**

- Measuring thickness of layers:
  - PANI layer by spin-coating 70nm
  - PANI layer by drop-casting 4um
  - PEDOT:PSS layer 70 nm
  - PEI layer 70nm
  - Mg4Ni layer & Pt layer 200nm



Clockwise from top left, PANI by spin-coating, PEDOT:PSS, PANI by drop-casting, and PEI.

# CHARACTERIZATION - N&K SPECTROPHOMETER

- Measuring refractive index and reflectivity of layers.
  - PEDOT:PSS layer
    - 1.68 as refractive index
    - About 94% as reflectivity @ 900nm
  - PEI layer
    - 1.62 as refractive index
    - About 92% as reflectivity @ 900nm



N&K spectrophotometer

# CHARACTERIZATION - N&K

#### SPECTROPHOMETER

- Measuring refractive index and reflectivity of layers.
  - PEDOT:PSS layer
    - 1.68 as refractive index
    - About 94% as reflectivity @ 900nm
  - PEI layer
    - 1.62 as refractive index
    - About 92% as reflectivity @ 900nm



Clockwise from top left, index of refraction and reflectivity of PEDOT:PSS and reflectivity and index of refraction of PEI layer.

# **CHARACTERIZATION - AFM**

Material	Thickness	RMS Roughness (nm)		
PEDOT:PSS spin-coat	~70 nm	7.59		
PEI spin-coat	~70 nm	Equal to PANI		
PANI spin-coat	~70 nm	12.36		
PANI drop-cast	~4 μm	130.16		
Mg4Ni/Pt sputtered	200 nm	4.67		
Glass	1.5 mm	5.27		



# RESULTS AND LOOKING FORWARD



Final prototype showing hydrogenated Mg<sub>4</sub>Ni (above) compared to its reflective state (below)

- Further Characterization
  - optical properties for modelling
  - N-and-K Spectrometry
  - More techniques and equations
  - Mechanism of hydrogen diffusion for electrical modelling
- Prototype Next Steps
  - Deposit thicker PEI
  - Thinner Mg<sub>4</sub>Ni
  - Smoother PANI
- Scale-up Considerations



# **Executive Committee**

- •Project Leader Jake Steiner
- •Secretary & Cinematographer Kari McPartland
- •Treasurer Glenn Pastel

# **Prototype Committee**

•Eshwari Murty •Kari McPartland

# **Design Committees**

**Optical Analysis** 

•Glenn Pastel

•Soo-Hwan Jang

•Ryan Tillman

**Electrical Analysis** 

•Jake Steiner

#### ACKNOWLEDGMENTS

Dr. Ray Phaneuf Dr. Rob Briber Dr. Ichiro Takeuchi Xin Zhang Sean Fackler Dr. Aldo Ponce Dr. Aldo Ponce Dr. Richard Kaner the Fablab staff the rest of the Materials Science Department faculty and staff and YOU, our supporters :D



- 1. Baetens R, Jelle BP, and A Gustaven. "Properties, requirements and possibilities of smart windows for dynamic daylight and solar energy control in buildings: A state-of-the-art review." Solar Energy Materials and Solar Cells, vol. 94; pp.87-105. 2010.
- 2. Tajima K, Hotta H et al. "Electrochromic switchable mirror glass fabricated using adhesive electrolyte layer." Applied Physics Letters, vol. 101. 2012.
- 3. Kirchmeyer S and K Reuter. "Scientific importance, properties, and growing applications of poly(3,4-ethylenedioxythiophene)." Journal of Materials Chemistry, vol. 15; pp. 2077-88. 2005.
- 4. Deepa M, Ahmad S, et al. "Electrochromic properties of polyaniline thin film nanostructures derived from solutions of ionic liquid/polyethylene glycol." Electrochimica Acta, vol. 52; pp. 7453-63. 2007.



#### FACILITIES

- Dr. Briber's lab to use stuffs such as glasswares and a spin-coater machine to synthesize polymer.
- Dr. Hu's lab To use the Meyer Rod deposition equipment.
- Sputter deposition machine in Fablab To fabricate the Mg<sub>4</sub>Ni/Pt.
- Dr. Takeuchi's lab To use EDS analyze the layer of Mg4Ni/Pt.
- Fablab in IREAP building To utilize an ellipsometer to measure indices of refraction for each layer.

- Fablab in KIM building To utilize a profilometer to characterize to find out the thickness of each layer exactly.
- Fablab in KIM building To utilize a spectrometer to characterize under various applied voltages.
- Dr. Phaneuf's lab To use COMSOL Multiphysics 4.4 to set proper models up.
- Dr Wuttig's lab Getting information of electrical properties requires a variety of machines.

# WORK PLAN

GANTT.	$\Rightarrow$		2014								
Name	Begin date	End date	Week 12 3/16	l Week 13 3/23	l Week 14 3/30	l Week 15 4/6	Week 16 4/13	l Week 17 4/20	l Week 18 4/27	l Week 19 5/4	Week 20 5/11
Documentation	2/10	5/6		_	_	_			D	ocumentation	
Planning & Research	2/10	3/25	Planning &	Research							
Prototype Design & Modelling	3/13	5/7							Prototype Des	sign & Modelling	
<ul> <li>Approximate Material Parameters</li> </ul>	3/13	3/26	proximate Material	Parameters							
Confirm Design & Predicted Prope	. 3/24	3/26	Design & Predicte	d Properties							
<ul> <li>Electrial Analysis (Applied Voltage</li> </ul>	3/27	4/11		Electrial /	Analysis (Applied V	oltage & Resistivity)					
<ul> <li>Optical Analysis (COMSOL)</li> </ul>	3/31	5/7							Optical Ana	alysis (COMSQL)	
<ul> <li>Begin Polymer Synthesis (PEDOT,</li> </ul>	. 3/24	4/25				Begin I	Polymer Synthesis	(PEDOT, PANI, PE	b +		
<ul> <li>FabLab Time (Sputtering)</li> </ul>	4/28	4/29						FabLab Time	(Sputtering)		
<ul> <li>3rd Report</li> </ul>	4/9	4/15					3rd Report				
Implementation & Analysis	4/17	5/15								Impleme	ntation & Analysis
<ul> <li>Wet-bench Processing (Adhesion)</li> </ul>	4/17	5/8							Wet-bench Pro	cessing (Adhesion)	
<ul> <li>Thermal Analysis</li> </ul>	4/24	5/1							Thermal Analysis		
Characterization (Spectrometer &	4/28	5/7						Characteri	zation (Spectrome	ter & Reliability)	
Feedback & Evaluation	4/30	5/9							F	eedback & Evaluatio	n
<ul> <li>Final Report</li> </ul>	5/7	5/15								i de la companya de l	Final Report
Presentation & Poster	5/7	5/15								Pre	sentation & Poster

# BUDGET

Item	Purpose/Details	Provider	Rate	Shipping	Allocation		
Mg Sputtering Target 3" x 1/4"	Mg4Ni film (1.88:1 ratio Mg:Ni)	AJA Interational	\$185	Yes	\$	185.00	
E-Beam Deposition onto PEDOT	Using FabLab for deposition	FabLab	\$70/hr	-	\$	210.00	
Use of Ni target	Mg₄Ni film	FabLab	\$70/1000Å	-	\$	70.00	
Use of Pt target	4nm Pt film	FabLab	\$70/1000Å	-	\$	10.00	
Optical Borosilicate Slides 43 x 50mm	Samples	Tedpella	\$26.95/35 pcs	Yes/\$10	\$	41.95	
Clevios PEDOT (PH 1000 grade)	PEDOT layer	Herceaus (PA)	\$155/100g	Yes/\$35	\$	190.00	
Ammonium Peroxydisulfate (98%)	PANI processing	Sigma Aldrich	\$35.60/100G	Yes/67.18	\$	102.78	
Polyvinyl Butyral (Butvar B-98?)	0.5g other adhesive in GBL	Sigma Aldrich	\$22.40/100G	Yes/67.18	\$	22.40	
Poly(ethyleneimine) solution	PEI aka 0.1ml adhesive electrolyte	Sigma Aldrich	\$38.50/100ML	Yes/67.18	\$	38.50	
Gamma-butyrolactone (GBL)	10ml. Tajima used PVB in GBL	Sigma Aldrich	\$19/25G	Yes/67.18	\$	19.00	
Aniline (ACS reagent >99.5%)	Material Processing? PANI?	TedPella	\$30/100ML	Yes/~\$15?	\$	45.00	
Hydrochloric Acid	PEI processing	UMD	\$8.91/1L	-	\$	8.91	
DMSO methyl sulfoxide	CM138 - PEDOT processing	UMD	\$13.70/500ML	-	\$	13.70	
Dialysis cassettes	for PANI processing	Dr. Briber - free	\$150/10pk	-	\$	-	
Wet-Bench Lab Space/Fume Hood	Need to provide details of use	Dr. Briber - free	\$0 for use	-	\$	-	
Ellipsometer and Spectometer	For Unknown Mat. Properties	UMD	-	-	\$	-	
COMSOL Multiphysics - Wave Optics	Optical Modelling and Analysis	Dr. Phaneuf's Lab	MSE Dep.	-	\$	-	
				Sum:	\$	957.24	
				Remaining:	S	42.76	

#### **MOTIVATION**

- Smart windows change optical properties from transparent to absorbing under stimulus
- By minimizing heat loss and gain they can thereby maximize building energy efficiency
- Japanese group led by Tajima created reflective smart window, better for deterring radiative heating
- ITO and WO<sub>3</sub> used in the original configuration limit financial viability
- We propose organic substitutes to reduce costs while maintaining effectiveness of the switchable mirror



From SwitchLite website: http://www.switchlite.com/home.html#



- Hard to estimate due to novel combination of layers
- PANI/ PEDOT:PSS interface could be problematic
- Affordability could be an issue if a Pt or Pd target cannot be found
- Potential for failure/frustration in both properly

#### BACKGROUND

- Most smart windows employ solidstate reactions via cation transport
- Cations diffuse in and out of materials like WO<sub>3</sub> causing structural changes and thus changes in optical properties
- Electrical potential induces diffusion of cations
- Devices need electrode/active layer/electrolyte/charge storage/electrode structure



#### **PREVIOUS WORK**

- The Baetens review specifies minimum performance criteria for commercialization
- WO<sub>3</sub> and ITO meet these requirements, but are expensive and hard to scale up
- Nb- Ni- and Ir- oxides have similar expense drawbacks
- PEDOT:PSS and PANI are electrochromatic polymers but may be UV sensitive
- Tajima's device uses Mg<sub>4</sub>Ni, which turns reflective when it takes in hydrogen, and PEI





- Incorporating PEDOT:PSS and PANI into the switchable mirror is a novel approach
- Opportunity to observe polymer interfaces
- New processing approaches
- Sputtering of Mg<sub>4</sub>Ni/Pt on polymer substrate
- Hydrogen diffusion modeling across Mg<sub>4</sub>Ni, PEI, and PANI
- Stress analysis in order to improve device lifetime despite thermal cycling

# BROADER IMPACT



• Windows are some of the least efficient building components

 Reduce dependence on rare earth elements like indium in ITO

 Smart windows are in response to stricter building energy regulations and a refocus on sustainability

 Lowering energy costs of buildings also reduces their environmental impact

 Anywhere windows are used: houses, factories, cars, planes

# **ETHICAL ISSUES**

- Actual device materials are non-toxic, no human health concern
- Chemicals in synthesis must be handled properly
- Magnesium reacts violently to H<sub>2</sub>O at room temperature
- HCl and aniline in PANI synthesis are also dangerous
- Potential success could lead to improvement of sustainable building design

